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First Example of 1,3-Dipolar Cycloaddition of Carbonyl Ylides to Cyclopropenes

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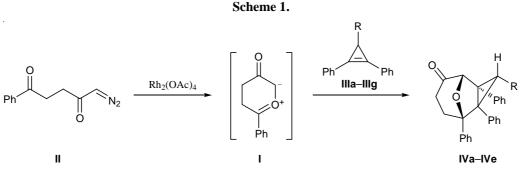
Abstract—Carbonyl ylide generated from 1-diazo-5-phenylpentane-2,5-dione in the presence of $Rh_2(OAc)_4$ reacts with 3-substituted cyclopropenes following the 1,3-dipolar cycloaddition pattern to afford substituted 9-oxatricyclo[3.3.1.0^{2,4}]nonan-6-ones.

1,3-Dipolar cycloaddition of carbonyl ylides, which are generated from diazo compounds in the presence of metal complex catalysts, to unsaturated compounds can be regarded as a promising synthetic route to oxygen-containing heterocyclic systems [1]. Reactions of carbonyl ylides with cyclopropene derivatives have not been studied, although cyclopropenes are known to readily participate in cycloaddition reactions [2].

We have found that carbonyl ylide I generated from 1-diazo-5-phenylpentane-2,5-dione (II) in the presence of Rh₂(OAc)₄ reacts with 3-substituted 1,2-diphenylcyclopropenes IIIa–IIIe in methylene chloride at room temperature to give 9-oxatricyclo[$3.3.1.0^{2.4}$]nonan-6ones IVa–IVe (Scheme 1). The structure of ketones IVa–IVe was established on the basis of their elemental compositions and spectral data. In the IR spectra of IVa–IVe we observed an absorption band at 1745 cm⁻¹ which belongs to the carbonyl group. Their ¹H NMR spectra contained a singlet at δ 4.58– 4.81 ppm from the 5-H proton neighboring to the epoxy bridge and multiplet signals at δ 2.09–3.35 ppm

from the CH₂CH₂ group, which are typical of 1.3-dipolar cycloaddition products derived from carbonyl ylide I [3]. Signals from protons in the three-membered ring appeared in the δ range from 2.1 to 3.3 ppm. In the ¹H NMR spectrum of adduct IVa, the *endo*proton gives a doublet at δ 2.09 ppm (J = 5 Hz), and the *exo*-proton signal is displaced upfield to δ 1.16 ppm (*J* = 5 Hz). The downfield position of the endo-proton signal results from deshielding effects of the two *trans*-arranged phenyl groups and the bridging oxygen atom which is located syn. In the ¹³C NMR spectra, signals from the carbon atoms attached to the bridging oxygen atom are located at $\delta_{\rm C}$ 85 ppm, carbon atoms of the ethylene moiety give signals at about $\delta_{\rm C}$ 34 ppm, the carbonyl carbon signal appears at $\delta_{\rm C}$ 205 ppm, and the other carbon signals occupy the $\delta_{\rm C}$ range from 38 to 48 ppm. The structure of **IVd** was proved by X-ray analysis (see figure).

On the other hand, carbonyl ylide I failed to react with cyclopropenes **IIIf** and **IIIg** having electronacceptor groups in position 3. It is known that carbonyl



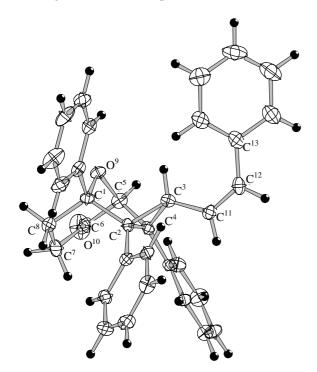
III, IV, R = H(a), Me (b), CH=CH₂ (c), (Z)-CH=CHPh (d), Ph (e), COOMe (f), CN (g).

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ylides as dipoles are capable of reacting with both electron-donor and electron-acceptor dipolarophiles [3]. Presumably, in our case the lowest unoccupied molecular orbital (LUMO) of the ylide interacts with the highest occupied molecular orbital (HOMO) of cyclopropene, i.e., ylide I acts as acceptor. According to the results of quantum-chemical calculations, an electron-donor substituent in position 3 of cyclopropene destabilizes this reagent and enhances its reactivity toward butadiene in the Diels-Alder reaction, while an electron-acceptor substituent in the same position stabilizes cyclopropene via considerable reduction of the HOMO energy; therefore, its reactivity becomes lower [4]. The formation of adducts IV as a single stereoisomer indicates exo-anti approach of carbonyl ylide I to cyclopropene and concerted mechanism of the addition.

EXPERIMENTAL

The IR spectra were recorded on a UR-20 spectrophotometer from 2% solutions in CHCl₃. The ¹H and ¹³C NMR spectra were measured on a Bruker DPX-300 instrument at 300.13 and 75.47 MHz, respectively, from solutions in CDCl₃. The purity of products was checked, and the reaction mixtures were analyzed, by TLC using Silufol UV-254 plates.



Structure of the molecule of rel-(1R,2R,3S,4S,5S)-1,2,4-triphenyl-3-[(Z)-2-phenylethenyl]-9-oxatricyclo[3.3.1.0^{2.4}]-nonan-6-one (**IVd**) according to the X-ray diffraction data.

Cyclopropenes **IIIa–IIIg** were synthesized by known methods [5].

rel-(1R,2R,4S,5S)-1,2,4-Triphenyl-9-oxatricyclo-[3.3.1.0^{2,4}]nonan-6-one (IVa). To a solution of 144 mg (0.76 mmol) of diazo ketone II and 175 mg (1.02 mmol) of 1,2-diphenylcyclopropene IIIa in 5 ml of anhydrous methylene chloride we added under stirring at room temperature in a stream of dry argon 2 mg of $Rh_2(OAc)_4$. The mixture was stirred for 1 h, the solvent was evaporated, and the residue was subjected to column chromatography on silica gel (gradient elution with hexane-ethyl acetate mixtures) to isolate 201 mg (75%) of ketone IVa, mp 198-201°C. ¹H NMR spectrum, δ , ppm: 1.16 d (1H, J = 5 Hz), 2.09 d (1H, J = 5 Hz), 2.25 d.d.d (1H, J = 14, 11, 8 Hz), 2.59 d.d (1H, J = 16, 8 Hz), 2.80 d.d (1H, J = 14, 9 Hz), 2.97 d.d.d (1H, J = 16, 11, 9 Hz), 4.81 s (1H), 6.78–6.87 (2H), 7.03–7.41 (13H). ¹³C NMR spectrum, δ_C, ppm: 23.3, 34.2, 34.5, 38.1, 45.5, 84.5. 85.5, 126.7, 126.6, 127.2, 127.8, 128.2, 128.3, 128.5, 128.6, 132.6, 134.2, 136.7, 140.3, 206.4. IR spectrum, v, cm⁻¹: 920, 990, 1040 s, 1080, 1135, 1265, 1445, 1500, 1600, 1740 v.s, 2965, 3070. Found, %: C 85.18; H 6.13. C₂₆H₂₂O₂. Calculated, %: C 85.22; H 6.05.

Ketones **IVe–IVd** were synthesized in a similar way.

rel-(1S,2S,3R,4R,5R)-3-Methyl-1,2,4-triphenyl-9oxatricyclo[3.3.1.0^{2,4}]nonan-6-one (IVb) was synthesized from 220 mg (1.16 mmol) of diazo ketone II and 300 mg (1.46 mmol) of 3-methyl-1,2-diphenylcyclopropene. Yield 345 mg (81%), mp 126-128°C. ¹H NMR spectrum, δ , ppm: 0.75 d (3H, J = 6 Hz), 2.13 q (1H, *J* = 6 Hz), 2.19 d.d.d (1H, *J* = 13, 9, 8 Hz), 2.50 d.d (1H, J = 17, 8 Hz), 2.78 d.d (1H, J = 13, 9 Hz), 2.93 d.t (1H, J = 17, 9 Hz), 4.81 s (1H), 6.90– 7.42 (15H). ¹³C NMR spectrum, δ_C , ppm: 11.2, 21.4, 33.9, 34.2, 42.1, 46.4, 85.6, 86.1, 126.0, 126.9, 127.8, 128.0, 128.1, 128.2, 128.3, 129.8, 131.9, 133.9, 134.0, 140.2, 205.8. IR spectrum, v, cm⁻¹: 1030, 1075, 1265, 1445, 1500, 1600, 1740 v.s, 2870, 2930, 2960, 3050. Found, %: C 85.44; H 6.89. C₂₇H₂₄O₂. Calculated, %: C 85.23; H 6.36.

rel-(1*R*,2*R*,3*S*,4*S*,5*S*)-1,2,4-Triphenyl-3-vinyl-9oxatricyclo[3.3.1.0^{2,4}]nonan-6-one (IVc) was synthesized from 228 mg (1.21 mmol) of diazo ketone II and 320 mg (1.47 mmol) of 3-vinyl-1,2-diphenylcyclopropene. Yield 312 mg (68%), mp 134–136°C. ¹H NMR spectrum, δ , ppm: 2.24 d.d.d (1H, *J* = 14, 9, 8 Hz), 2.59 d.d (1H, *J* = 17, 8 Hz), 2.79 d (1H, *J* = 10 Hz), 2.86 d.d (1H, *J* = 14, 9 Hz), 3.03 d.t (1H, *J* = 17, 9 Hz), 4.79 s (1H), 4.88 d.d (1H, *J* = 10, 2 Hz), 5.03 d.t (1H, J = 17, 10 Hz), 5.18 d.d (1H, J = 17, 2 Hz), 6.95–7.02 (2H), 7.08–7.11 (2H), 7.18–7.40 (11H). ¹³C NMR spectrum, $\delta_{\rm C}$, ppm: 31.8, 33.4, 34.2, 43.6, 46.8, 86.0, 86.4, 117.0, 125.8, 127.4, 127.9, 128.1, 128.2, 128.6, 130.0, 131.8, 133.3, 134.0, 134.3, 139.9, 205.0. IR spectrum, v, cm⁻¹: 915, 1060 s, 1080, 1140, 1445, 1500, 1600, 1740 v.s, 2965, 3065. Found, %: C 85.84; H 6.12. C₂₈H₂₄O₂. Calculated, %: C 85.68; H 6.16.

rel-(1R,2R,3S,4S,5S)-1,2,4-Triphenyl-3-[(Z)-2phenylethenyl]-9-oxatricyclo[3.3.1.0^{2,4}]nonan-6-one (IVd) was synthesized from 107 mg (0.57 mmol) of diazo ketone II and 200 mg (0.68 mmol) of 3-[(Z)-2phenylethenyl]-1,2-diphenylcyclopropene. Yield 179 mg (69%), mp 205–206°C. ¹H NMR spectrum, δ , ppm: 2.25 d.d.d (1H, J = 14, 10, 8 Hz), 2.59 d.d (1H, J = 17, 8 Hz), 2.88 d.d (1H, J = 14, 9 Hz), 3.02 d.d.d (1H, J =17, 10, 9 Hz), 3.23 d (1H, J = 11 Hz), 4.75 s (1H), 4.94 t (1H, J = 11 Hz), 6.32 d (1H, J = 11 Hz), 7.10-7.13 (2H), 7.18–7.44 (16H). 13 C NMR spectrum, δ_{C} , ppm: 27.2, 34.1, 34.3, 44.5, 46.8, 85.7, 86.5, 126.1, 127.1, 127.3, 127.5, 128.1, 128.4, 128.7, 129.3, 130.2, 130.9, 131.9, 133.0, 131.9, 133.0, 134.0, 137.3, 139.9, 204.9. IR spectrum, v, cm⁻¹: 1035, 1060, 1080, 1255, 1450, 1500, 1600, 1740 v.s, 2960, 3065. Found, %: C 87.10; H 5.95. C₃₄H₂₈O₂. Calculated, %: C 86.85; H 5.92.

rel-(1R,2R,3S,4S,5S)-1,2,3,4-Tetraphenyl-9-oxatricyclo[3.3.1.0^{2,4}]nonan-6-one (IVe) was synthesized from 99 mg (0.52 mmol) of diazo ketone II and 168 mg (0.63 mmol) of 1,2,3-triphenylcyclopropene. Yield 147 mg (66%), mp 174–177°C. ¹H NMR spectrum, δ, ppm: 2.26–2.37 (1H), 2.70–2.79 (1H), 2.94– 3.02 (1H), 3.23-3.35 (2H), 4.58 s (1H), 6.23-6.26 (2H), 6.77–7.33 (18H). ¹³C NMR spectrum, δ_C , ppm: 31.3, 32.4, 34.1, 46.3, 48.6, 61.8, 87.1, 88.3, 125.7, 126.0, 127.0, 127.96, 128.04, 128.06, 128.8, 129.4, 130.9, 131.0, 131.5, 131.8, 134.5, 135.8, 140.1, 204.9. IR spectrum, cm⁻¹: 920, 940, 980, 1020, 1040, 1060, 1080, 1140, 1180, 1240, 1310, 1350, 1420, 1450, 1500 br. 1600, 1720 v.s. 2860, 2880, 2940, 2960, 3040, 3070, 3090. Found, %: C 86.55; H 6.27. C₃₂H₂₆O₂. Calculated, %: C 86.85; H 5.92.

X-Ray analysis of compound IVd. Formula $C_{34}H_{28}O_2$; *M* 468.56; triclinic srystals, space group

*P*₁2₁/*c*₁ (no. 14); unit cell parameters: *a* = 12.627(2), *b* = 20.536(3), *c* = 19.097(3) Å; *α* = 90.00, *β* = 90.00, *γ* = 90.87°; *V* = 4951.43(130) Å³; *Z* = 8; *d*_{calc} = 1.257 g/cm³; *μ* = 0.078 mm⁻¹; *F*(000) = 912, MoK_α radiation, *λ* = 0.71073 Å, graphite monochromator. Below are given selected bond lengths (Å) and bond angles (deg): O⁹-C⁵ 1.439(2), O⁹-C¹ 1.444(3), O¹⁰-C⁶ 1.208(2), C¹-C⁸ 1.534(2), C¹-C² 1.553(3), C²-C⁴ 1.523(2), C²-C³ 1.528.2(2), C³-C¹¹ 1.472(2), C³-C⁴ 1.546(2), C⁴-C⁵ 1.532(2), C⁵-C⁶ 1.516(2), C⁶-C⁷ 1.503(3), C⁷-C⁸ 1.545(2), C¹¹-C¹² 1.335(2); C⁵O⁹C¹ 105.21(10), O⁹C¹C⁸ 106.57(11), O⁹C¹C² 103.06(10), C⁴C²C³ 60.87(8), C⁴C²C¹ 104.51(10), C³C²C¹ 112.54(11), C¹¹C³C² 123.60(12), C¹¹C³C⁴ 119.87(12), C²C³C⁴ 59.40(8), C²C⁴C⁵ 104.26(11). The complete crystallographic data set was deposited to the Cambridge Structural Database (CCDC).

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REFERENCES

- 1. Padwa, A. and Weingarten, M.D., *Chem. Rev.*, 1996, vol. 96, p. 223.
- Baird, M.S., *Top. Curr. Chem.*, 1987, vol. 144, p. 137; *Methods of Organic Chemistry*, de Meijere, A., Ed., Stuttgart: Georg Thieme, 1997, vol. 17d.
- Padwa, A., Fryxell, G., and Zhi, L., J. Org. Chem., 1990, vol. 112, p. 3100.
- 4. Xidos, J.L., Gosse, T.L., Burke, E.D., Poirier, R.A., and Burnell, D., J. Am. Chem. Soc., 2001, vol. 123, p. 5482.
- Longone, D.I. and Stehouwer, D.M., *Tetrahedron Lett.*, 1970, p. 1017; Fiato, R.A., Williams, J.B., and Battiste, M.A., *Synthesis*, 1974, p. 273; Padwa, A., Blacklook, T.J., Getman, D., Hatanaka, N., and Loza, R., *J. Org. Chem.*, 1978, vol. 43, p. 1481; Hughes, R.P., Trujillo, H.A., Egan, J.W., and Rheingold, A.L., *J. Am. Chem. Soc.*, 2000, vol. 122, p. 2261; Komendantov, M.I., Logosh, I.B., and Domnin, I.N., *Zh. Org. Khim.*, 1985, vol. 21, p. 1026; Breslow, R. and Chang, H.W., *J. Am. Chem. Soc.*, 1961, vol. 83, p. 2367; Breslow, R., Winter, R., and Battiste, M., *J. Org. Chem.*, 1959, vol. 24, p. 415; White, E.H., Winter, R.E.K., Graeve, R., Zirngibl, U., Friend, Maskill, M., Mende, U., Kreiling, G., Reisenauer, H.P., and Maier, G., *Chem. Ber.*, 1981, vol. 114, p. 3906.