# First Example of 1,3-Dipolar Cycloaddition of Carbonyl Ylides to Cyclopropenes 

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#### Abstract

Carbonyl ylide generated from 1-diazo-5-phenylpentane-2,5-dione in the presence of $\mathrm{Rh}_{2}(\mathrm{OAc})_{4}$ reacts with 3 -substituted cyclopropenes following the 1,3-dipolar cycloaddition pattern to afford substituted 9 -oxatricyclo[3.3.1.0 ${ }^{2,4}$ ]nonan-6-ones.


1,3-Dipolar cycloaddition of carbonyl ylides, which are generated from diazo compounds in the presence of metal complex catalysts, to unsaturated compounds can be regarded as a promising synthetic route to oxygen-containing heterocyclic systems [1]. Reactions of carbonyl ylides with cyclopropene derivatives have not been studied, although cyclopropenes are known to readily participate in cycloaddition reactions [2].

We have found that carbonyl ylide I generated from 1-diazo-5-phenylpentane-2,5-dione (II) in the presence of $\mathrm{Rh}_{2}(\mathrm{OAc})_{4}$ reacts with 3 -substituted 1,2-diphenylcyclopropenes IIIa-IIIe in methylene chloride at room temperature to give 9 -oxatricyclo[3.3.1.0 $\left.{ }^{2,4}\right]$ nonan-6ones IVa-IVe (Scheme 1). The structure of ketones IVa-IVe was established on the basis of their elemental compositions and spectral data. In the IR spectra of IVa-IVe we observed an absorption band at $1745 \mathrm{~cm}^{-1}$ which belongs to the carbonyl group. Their ${ }^{1} \mathrm{H}$ NMR spectra contained a singlet at $\delta$ 4.584.81 ppm from the $5-\mathrm{H}$ proton neighboring to the epoxy bridge and multiplet signals at $\delta 2.09-3.35 \mathrm{ppm}$
from the $\mathrm{CH}_{2} \mathrm{CH}_{2}$ group, which are typical of 1,3-dipolar cycloaddition products derived from carbonyl ylide I [3]. Signals from protons in the three-membered ring appeared in the $\delta$ range from 2.1 to 3.3 ppm . In the ${ }^{1} \mathrm{H}$ NMR spectrum of adduct IVa, the endoproton gives a doublet at $\delta 2.09 \mathrm{ppm}(J=5 \mathrm{~Hz})$, and the exo-proton signal is displaced upfield to $\delta 1.16 \mathrm{ppm}(J=5 \mathrm{~Hz})$. The downfield position of the endo-proton signal results from deshielding effects of the two trans-arranged phenyl groups and the bridging oxygen atom which is located syn. In the ${ }^{13} \mathrm{C}$ NMR spectra, signals from the carbon atoms attached to the bridging oxygen atom are located at $\delta_{\mathrm{C}} 85 \mathrm{ppm}$, carbon atoms of the ethylene moiety give signals at about $\delta_{\mathrm{C}} 34 \mathrm{ppm}$, the carbonyl carbon signal appears at $\delta_{\mathrm{C}} 205 \mathrm{ppm}$, and the other carbon signals occupy the $\delta_{\mathrm{C}}$ range from 38 to 48 ppm . The structure of IVd was proved by X-ray analysis (see figure).

On the other hand, carbonyl ylide $\mathbf{I}$ failed to react with cyclopropenes IIIf and IIIg having electronacceptor groups in position 3. It is known that carbonyl

## Scheme 1.



III, IV, $\mathrm{R}=\mathrm{H}(\mathbf{a}), \mathrm{Me}(\mathbf{b}), \mathrm{CH}=\mathrm{CH}_{2}(\mathbf{c}),(\mathrm{Z})-\mathrm{CH}=\mathrm{CHPh}(\mathbf{d}), \mathrm{Ph}(\mathbf{e}), \mathrm{COOMe}(\mathbf{f}), \mathrm{CN}(\mathbf{g})$.
ylides as dipoles are capable of reacting with both electron-donor and electron-acceptor dipolarophiles [3]. Presumably, in our case the lowest unoccupied molecular orbital (LUMO) of the ylide interacts with the highest occupied molecular orbital (HOMO) of cyclopropene, i.e., ylide I acts as acceptor. According to the results of quantum-chemical calculations, an electron-donor substituent in position 3 of cyclopropene destabilizes this reagent and enhances its reactivity toward butadiene in the Diels-Alder reaction, while an electron-acceptor substituent in the same position stabilizes cyclopropene via considerable reduction of the HOMO energy; therefore, its reactivity becomes lower [4]. The formation of adducts IV as a single stereoisomer indicates exo-anti approach of carbonyl ylide I to cyclopropene and concerted mechanism of the addition.

## EXPERIMENTAL

The IR spectra were recorded on a UR-20 spectrophotometer from $2 \%$ solutions in $\mathrm{CHCl}_{3}$. The ${ }^{1} \mathrm{H}$ and ${ }^{13} \mathrm{C}$ NMR spectra were measured on a Bruker DPX300 instrument at 300.13 and 75.47 MHz , respectively, from solutions in $\mathrm{CDCl}_{3}$. The purity of products was checked, and the reaction mixtures were analyzed, by TLC using Silufol UV-254 plates.


Structure of the molecule of rel-( $1 R, 2 R, 3 S, 4 S, 5 S$ )-1,2,4-triphenyl-3-[(Z)-2-phenylethenyl]-9-oxatricyclo[3.3.1.0 $\left.{ }^{2,4}\right]$ -nonan-6-one (IVd) according to the X-ray diffraction data.

Cyclopropenes IIIa-IIIg were synthesized by known methods [5].
rel-(1R,2R,4S,5S)-1,2,4-Triphenyl-9-oxatricyclo[3.3.1.0 ${ }^{2,4}$ ]nonan-6-one (IVa). To a solution of $144 \mathrm{mg}(0.76 \mathrm{mmol})$ of diazo ketone II and 175 mg ( 1.02 mmol ) of 1,2 -diphenylcyclopropene IIIa in 5 ml of anhydrous methylene chloride we added under stirring at room temperature in a stream of dry argon 2 mg of $\mathrm{Rh}_{2}(\mathrm{OAc})_{4}$. The mixture was stirred for 1 h , the solvent was evaporated, and the residue was subjected to column chromatography on silica gel (gradient elution with hexane-ethyl acetate mixtures) to isolate 201 mg ( $75 \%$ ) of ketone IVa, mp 198$201{ }^{\circ} \mathrm{C}$. ${ }^{1} \mathrm{H}$ NMR spectrum, $\delta, \mathrm{ppm}: 1.16 \mathrm{~d}(1 \mathrm{H}, J=$ $5 \mathrm{~Hz}), 2.09 \mathrm{~d}(1 \mathrm{H}, J=5 \mathrm{~Hz}), 2.25$ d.d.d $(1 \mathrm{H}, J=14$, $11,8 \mathrm{~Hz}), 2.59$ d.d $(1 \mathrm{H}, J=16,8 \mathrm{~Hz}), 2.80 \mathrm{~d} . \mathrm{d}(1 \mathrm{H}$, $J=14,9 \mathrm{~Hz}$ ), 2.97 d.d.d ( $1 \mathrm{H}, J=16,11,9 \mathrm{~Hz}$ ), 4.81 s $(1 \mathrm{H}), 6.78-6.87(2 \mathrm{H}), 7.03-7.41(13 \mathrm{H}) .{ }^{13} \mathrm{C}$ NMR spectrum, $\delta_{\mathrm{C}}$, ppm: $23.3,34.2,34.5,38.1,45.5,84.5$, 85.5, 126.7, 126.6, 127.2, 127.8, 128.2, 128.3, 128.5, 128.6, 132.6, 134.2, 136.7, 140.3, 206.4. IR spectrum, $v, \mathrm{~cm}^{-1}: 920,990,1040 \mathrm{~s}, 1080,1135,1265,1445$, $1500,1600,1740$ v.s, 2965, 3070. Found, \%: C 85.18; H 6.13. $\mathrm{C}_{26} \mathrm{H}_{22} \mathrm{O}_{2}$. Calculated, \%: C 85.22; H 6.05 .

Ketones IVe-IVd were synthesized in a similar way.
rel-(1S,2S,3R,4R,5R)-3-Methyl-1,2,4-triphenyl-9oxatricyclo $\left[\right.$ 3.3.1. $0^{2,4}$ ]nonan-6-one (IVb) was synthesized from $220 \mathrm{mg}(1.16 \mathrm{mmol})$ of diazo ketone II and 300 mg ( 1.46 mmol ) of 3-methyl-1,2-diphenylcyclopropene. Yield 345 mg ( $81 \%$ ), $\mathrm{mp} 126-128^{\circ} \mathrm{C}$. ${ }^{1} \mathrm{H}$ NMR spectrum, $\delta, \mathrm{ppm}: 0.75 \mathrm{~d}(3 \mathrm{H}, J=6 \mathrm{~Hz})$, $2.13 \mathrm{q}(1 \mathrm{H}, J=6 \mathrm{~Hz}), 2.19$ d.d.d $(1 \mathrm{H}, J=13,9,8 \mathrm{~Hz})$, 2.50 d.d $(1 \mathrm{H}, J=17,8 \mathrm{~Hz}), 2.78$ d.d $(1 \mathrm{H}, J=13$, $9 \mathrm{~Hz}), 2.93$ d.t $(1 \mathrm{H}, J=17,9 \mathrm{~Hz}), 4.81 \mathrm{~s}(1 \mathrm{H}), 6.90-$ $7.42(15 \mathrm{H}) .{ }^{13} \mathrm{C}$ NMR spectrum, $\delta_{\mathrm{C}}$, ppm: 11.2, 21.4, 33.9, 34.2, 42.1, 46.4, 85.6, 86.1, 126.0, 126.9, 127.8, $128.0,128.1,128.2,128.3,129.8,131.9,133.9,134.0$, 140.2, 205.8. IR spectrum, $v, \mathrm{~cm}^{-1}: 1030,1075,1265$, 1445, 1500, 1600, 1740 v.s, 2870, 2930, 2960, 3050. Found, \%: C 85.44; H 6.89. $\mathrm{C}_{27} \mathrm{H}_{24} \mathrm{O}_{2}$. Calculated, \%: C 85.23; H 6.36 .
rel-(1R,2R,3S,4S,5S)-1,2,4-Triphenyl-3-vinyl-9oxatricyclo[3.3.1.0 ${ }^{2,4}$ ]nonan-6-one (IVc) was synthesized from $228 \mathrm{mg}(1.21 \mathrm{mmol})$ of diazo ketone $\mathbf{I I}$ and 320 mg ( 1.47 mmol ) of 3 -vinyl-1,2-diphenylcyclopropene. Yield 312 mg ( $68 \%$ ), $\mathrm{mp} 134-136^{\circ} \mathrm{C}$. ${ }^{1} \mathrm{H}$ NMR spectrum, $\delta$, ppm: 2.24 d.d.d $(1 \mathrm{H}, J=14,9$, $8 \mathrm{~Hz}), 2.59 \mathrm{~d} . \mathrm{d}(1 \mathrm{H}, J=17,8 \mathrm{~Hz}), 2.79 \mathrm{~d}(1 \mathrm{H}$, $J=10 \mathrm{~Hz}), 2.86$ d.d $(1 \mathrm{H}, J=14,9 \mathrm{~Hz}), 3.03$ d.t $(1 \mathrm{H}$, $J=17,9 \mathrm{~Hz}), 4.79 \mathrm{~s}(1 \mathrm{H}), 4.88 \mathrm{~d} . \mathrm{d}(1 \mathrm{H}, J=10,2 \mathrm{~Hz})$,
5.03 d.t $(1 \mathrm{H}, J=17,10 \mathrm{~Hz}), 5.18$ d.d $(1 \mathrm{H}, J=17$, $2 \mathrm{~Hz}), 6.95-7.02(2 \mathrm{H}), 7.08-7.11(2 \mathrm{H}), 7.18-7.40$ $(11 \mathrm{H}) .{ }^{13} \mathrm{C}$ NMR spectrum, $\delta_{\mathrm{C}}$, ppm: $31.8,33.4,34.2$, $43.6,46.8,86.0,86.4,117.0,125.8,127.4,127.9$, 128.1, 128.2, 128.6, 130.0, 131.8, 133.3, 134.0, 134.3, 139.9, 205.0. IR spectrum, $v, \mathrm{~cm}^{-1}: 915,1060 \mathrm{~s}, 1080$, 1140, 1445, 1500, 1600, 1740 v.s, 2965, 3065. Found, \%: C 85.84; H 6.12. $\mathrm{C}_{28} \mathrm{H}_{24} \mathrm{O}_{2}$. Calculated, \%: C 85.68; H 6.16.
rel-(1R,2R,3S,4S,5S)-1,2,4-Triphenyl-3-[(Z)-2-phenylethenyl]-9-oxatricyclo[3.3.1.0 ${ }^{2,4}$ ]nonan-6-one (IVd) was synthesized from $107 \mathrm{mg}(0.57 \mathrm{mmol})$ of diazo ketone II and $200 \mathrm{mg}(0.68 \mathrm{mmol})$ of $3-[(Z)-2-$ phenylethenyl]-1,2-diphenylcyclopropene. Yield 179 mg (69\%), mp 205-206 ${ }^{\circ} \mathrm{C} .{ }^{1} \mathrm{H}$ NMR spectrum, $\delta$, ppm: 2.25 d.d.d $(1 \mathrm{H}, J=14,10,8 \mathrm{~Hz}), 2.59$ d.d $(1 \mathrm{H}, J=17$, $8 \mathrm{~Hz}), 2.88 \mathrm{~d} . \mathrm{d}(1 \mathrm{H}, J=14,9 \mathrm{~Hz}), 3.02$ d.d.d $(1 \mathrm{H}, J=$ $17,10,9 \mathrm{~Hz}), 3.23 \mathrm{~d}(1 \mathrm{H}, J=11 \mathrm{~Hz}), 4.75 \mathrm{~s}(1 \mathrm{H})$, $4.94 \mathrm{t}(1 \mathrm{H}, J=11 \mathrm{~Hz}), 6.32 \mathrm{~d}(1 \mathrm{H}, J=11 \mathrm{~Hz}), 7.10-$ $7.13(2 \mathrm{H}), 7.18-7.44(16 \mathrm{H}) .{ }^{13} \mathrm{C}$ NMR spectrum, $\delta_{\mathrm{C}}$, ppm: 27.2, 34.1, 34.3, 44.5, 46.8, 85.7, 86.5, 126.1, 127.1, 127.3, 127.5, 128.1, 128.4, 128.7, 129.3, 130.2, $130.9,131.9,133.0,131.9,133.0,134.0,137.3,139.9$, 204.9. IR spectrum, $v, \mathrm{~cm}^{-1}: 1035,1060,1080,1255$, 1450, 1500, 1600, 1740 v.s, 2960, 3065. Found, \%: C 87.10; H 5.95. $\mathrm{C}_{34} \mathrm{H}_{28} \mathrm{O}_{2}$. Calculated, \%: С 86.85; H 5.92.
rel-(1R,2R,3S,4S,5S)-1,2,3,4-Tetraphenyl-9-oxatricyclo[3.3.1.0 ${ }^{2,4}$ ]nonan-6-one (IVe) was synthesized from $99 \mathrm{mg}(0.52 \mathrm{mmol})$ of diazo ketone II and 168 mg ( 0.63 mmol ) of 1,2,3-triphenylcyclopropene. Yield $147 \mathrm{mg}(66 \%)$, mp $174-177^{\circ} \mathrm{C}$. ${ }^{1} \mathrm{H}$ NMR spectrum, $\delta$, ppm: 2.26-2.37 (1H), 2.70-2.79 (1H), 2.94$3.02(1 \mathrm{H}), 3.23-3.35(2 \mathrm{H}), 4.58 \mathrm{~s}(1 \mathrm{H}), 6.23-6.26$ $(2 \mathrm{H}), 6.77-7.33(18 \mathrm{H}) .{ }^{13} \mathrm{C}$ NMR spectrum, $\delta_{\mathrm{C}}$, ppm: $31.3,32.4,34.1,46.3,48.6,61.8,87.1,88.3,125.7$, $126.0,127.0,127.96,128.04,128.06,128.8,129.4$, $130.9,131.0,131.5,131.8,134.5,135.8,140.1,204.9$. IR spectrum, $\mathrm{cm}^{-1}: 920,940,980,1020,1040,1060$, 1080, 1140, 1180, 1240, 1310, 1350, 1420, 1450, 1500 br, 1600, 1720 v.s, 2860, 2880, 2940, 2960, 3040, 3070, 3090. Found, \%: C 86.55; H 6.27. $\mathrm{C}_{32} \mathrm{H}_{26} \mathrm{O}_{2}$. Calculated, \%: C 86.85; H 5.92.

X-Ray analysis of compound IVd. Formula $\mathrm{C}_{34} \mathrm{H}_{28} \mathrm{O}_{2} ; M 468.56$; triclinic srystals, space group
$P_{1} 2_{1} / c_{1}$ (no. 14); unit cell parameters: $a=12.627(2)$, $b=20.536(3), c=19.097(3) \AA ; \alpha=90.00, \beta=90.00$, $\gamma=90.87^{\circ} ; V=4951.43(130) \AA^{3} ; Z=8 ; d_{\text {calc }}=$ $1.257 \mathrm{~g} / \mathrm{cm}^{3} ; \mu=0.078 \mathrm{~mm}^{-1} ; F(000)=912, \operatorname{Mo} K_{\alpha}$ radiation, $\lambda=0.71073 \AA$, graphite monochromator. Below are given selected bond lengths ( $\AA$ ) and bond angles (deg): $\mathrm{O}^{9}-\mathrm{C}^{5} 1.439(2), \mathrm{O}^{9}-\mathrm{C}^{1} 1.444(3), \mathrm{O}^{10}-\mathrm{C}^{6}$ $1.208(2), \mathrm{C}^{1}-\mathrm{C}^{8} 1.534(2), \mathrm{C}^{1}-\mathrm{C}^{2} 1.553(3), \mathrm{C}^{2}-\mathrm{C}^{4}$ $1.523(2), C^{2}-C^{3} 1.528 .2(2), C^{3}-C^{11} 1.472(2), C^{3}-C^{4}$ $1.546(2), C^{4}-C^{5} 1.532(2), C^{5}-C^{6} 1.516(2), C^{6}-C^{7}$ $1.503(3), \mathrm{C}^{7}-\mathrm{C}^{8} 1.545(2), \mathrm{C}^{11}-\mathrm{C}^{12} 1.335(2) ; \mathrm{C}^{5} \mathrm{O}^{9} \mathrm{C}^{1}$ 105.21(10), $\mathrm{O}^{9} \mathrm{C}^{1} \mathrm{C}^{8} 106.57(11), \mathrm{O}^{9} \mathrm{C}^{1} \mathrm{C}^{2} 103.06(10)$, $C^{4} C^{2} C^{3}$ 60.87(8), $C^{4} C^{2} C^{1} 104.51(10), C^{3} C^{2} C^{1}$ 112.54(11), $\mathrm{C}^{11} \mathrm{C}^{3} \mathrm{C}^{2}$ 123.60(12), $\mathrm{C}^{11} \mathrm{C}^{3} \mathrm{C}^{4}$ 119.87(12), $C^{2} C^{3} C^{4} 59.40(8), C^{2} C^{4} C^{5}$ 104.26(11). The complete crystallographic data set was deposited to the Cambridge Structural Database (CCDC).

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## REFERENCES

1. Padwa, A. and Weingarten, M.D., Chem. Rev., 1996, vol. 96, p. 223.
2. Baird, M.S., Top. Curr. Chem., 1987, vol. 144, p. 137; Methods of Organic Chemistry, de Meijere, A., Ed., Stuttgart: Georg Thieme, 1997, vol. 17d.
3. Padwa, A., Fryxell, G., and Zhi, L., J. Org. Chem., 1990, vol. 112, p. 3100.
4. Xidos, J.L., Gosse, T.L., Burke, E.D., Poirier, R.A., and Burnell, D., J. Am. Chem. Soc., 2001, vol. 123, p. 5482.
5. Longone, D.I. and Stehouwer, D.M., Tetrahedron Lett., 1970, p. 1017; Fiato, R.A., Williams, J.B., and Battiste, M.A., Synthesis, 1974, p. 273; Padwa, A., Blacklook, T.J., Getman, D., Hatanaka, N., and Loza, R., J. Org. Chem., 1978, vol. 43, p. 1481; Hughes, R.P., Trujillo, H.A., Egan, J.W., and Rheingold, A.L., J. Am. Chem. Soc., 2000, vol. 122, p. 2261; Komendantov, M.I., Logosh, I.B., and Domnin, I.N., Zh. Org. Khim., 1985, vol. 21, p. 1026; Breslow, R. and Chang, H.W., J. Am. Chem. Soc., 1961, vol. 83, p. 2367; Breslow, R., Winter, R., and Battiste, M., J. Org. Chem., 1959, vol. 24, p. 415; White, E.H., Winter, R.E.K., Graeve, R., Zirngibl, U., Friend, Maskill, M., Mende, U., Kreiling, G., Reisenauer, H.P., and Maier, G., Chem. Ber., 1981, vol. 114, p. 3906.
